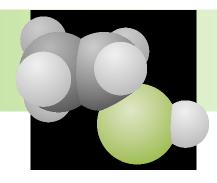
## **CHEMICALS**

**Project Fact Sheet** 

### CATALYST FOR CH<sub>4</sub>-CO CONVERSION



### **B**ENEFITS

- · Global economic benefits
- More efficient utilization of domestic and remote gas resources
- Decreased oil importation expenditures

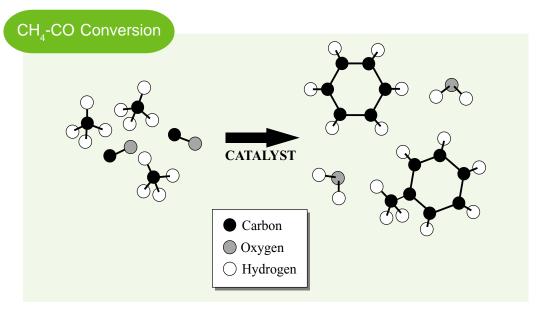
#### **APPLICATIONS**

This technology is designed for the direct conversion of natural gas to liquid petrochemicals and intermediates. Considering that mature markets exist for aromatics and their respective derivatives, industry-wide acceptance, endorsement, and commercial implementation of the process is expected.

# New technology allows for natural gas to be directly converted to liquid chemicals

High cost and inefficient processes have limited the feasibility of large-scale production of liquid chemicals from the U.S. domestic coal and natural gas supply. Current gas-to-liquid transformation involves an indirect route in which natural gas is first converted to synthesis gas and subsequently converted to liquids by available technologies such as the Fisher-Tropsch process. Up to 60 percent of the cost of this process is associated with the intermediate production and purification of synthesis gas. Consequently, there is considerable interest in the development of new, economical catalyst and process technologies that directly convert natural gas to liquids.

Project partners will develop a catalyst that allows for the direct conversion of methane to aromatics (benzene, toluene, xylene, and naphthalene) and light olefins. Earlier projects exploring the direct conversion of methane to aromatics were economically nonviable due to low methane conversion and poor selectivity. This process addresses these concerns. It is suitable for fixed bed reactor technology and is capable of being quickly scaled-up to commercial capacity. Successful completion of this project will provide a significant increase in the utilization of domestic and remote natural gas reserves because it directly converts natural gas to higher molecular weight products of substantially higher market value.



The diagram shows the process for the direct conversion of CH<sub>4</sub> and CO to aromatics and olefins.



### **Project Description**

**Goal**: The goal of this project is to develop a novel catalyst and process for the direct conversion of methane and carbon monoxide to aromatics and light olefins.

Conventional  $\mathrm{CH_4}$  dehydro-aromatization involves the direct conversion of  $\mathrm{CH_4}$  to benzene, ethylene and napthalene. The process suffers from low  $\mathrm{CH_4}$  conversion, poor selectivity, and thermodynamic constraints, which neccesitate operation at low  $\mathrm{CH_4}$  partial pressures and temperatures exceeding 973 K. Project partners are developing a multi-functional catalyst and process that overcomes the inherent limitations of  $\mathrm{CH_4}$  dehydro-aromatization and permits the direct conversion of  $\mathrm{CH_4}$  to aromatics and olefins at temperatures below 973 K and at pressures significantly above 1 atmosphere.

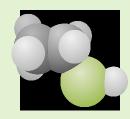
### **Progress and Milestones**

Significant progress has been made in the following areas:

- Demonstrated that a multi-functional catalyst can be prepared and activated for simultaneous CH<sub>3</sub>-CO conversion.
- Demonstrated that at least one catalyst composition exits such that CH<sub>4</sub> conversion and hydrocarbon selectivity for simultaneous CH<sub>4</sub>-CO conversion are better than those demonstrated for pure CH<sub>4</sub> dehydro-aromatization.
- Demonstrated that the multi-functional catalyst permits operation at reaction temperatures lower than 973 K and pressures significantly higher than 1 atm.

Future research focuses on achieving the following milestones:

- Increase the product selectivity of benzene, toluene, and xylene to 75 weight percent on a hydrocarbon basis at a CH<sub>4</sub> conversion of 25 percent.
- Demonstrate stable and continuous performance of the process for one week.



### PROJECT PARTNERS

CeraMem Corporation Waltham, MA

UOP, LLC Des Plaines, IL

Zeolyst International Valley Forge, PA

Dynasim Technical Services, Inc. Manhattan, KS

## FOR ADDITIONAL INFORMATION, PLEASE CONTACT:

Charles Russomanno
Office of Industrial Technologies
Phone: (202) 586-7543
Fax: (202) 586-3237
charles.russomanno@ee.doe.gov

Please send any comments, questions, or suggestions to webmaster.oit@ee.doe.gov

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Office of Industrial Technologies Energy Efficiency and Renewable Energy U.S. Department of Energy Washington, D.C. 20585

